

APPROACHES TO THE ANTIBIOTIC SPARSOMYCIN. AN EFFICIENT SYNTHESIS
OF THE CYSTEINOL MONO-OXODITHIOACETAL MOIETY

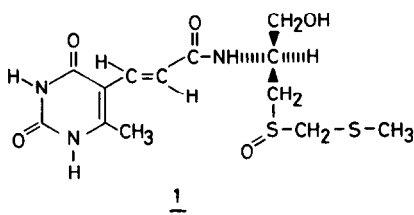
H.C.J. Ottenheijm^x and R.M.J. Liskamp

Department of Organic Chemistry, University of Nijmegen,
Toernooiveld, Nijmegen, The Netherlands

(Received in UK 19 April 1978; accepted for publication 12 May 1978)

Sparsomycin (1), a fermentation product¹ of *Streptomyces sparsogenes*, has attracted much attention because of its biological activity and its unique -S(O)CH₂-SCH₃ moiety. It displays a broad spectrum of in vitro activity against bacteria and shows antifungal activity². Its activity appears to be related to its ability to inhibit protein synthesis by blocking the ribosomal peptidyl transferase function³. In addition sparsomycin shows antitumor activity². Recently, the blocking of the peptidyl transferase function⁴ and antitumor activity⁵ have been studied with sparsomycin analogs in which the S(O)CH₂SCH₃ moiety had been replaced by more easily accessible side chains.

The structure 1 has been proposed by Wiley and MacKellar⁶. Recently, this structure has been substantiated by the synthesis of S-deoxo-sparsomycin by us⁷ and others^{4,5}. However, a synthesis of 1, including the mono-oxodithioacetal side chain in the cysteinol moiety, has not yet appeared in literature. We wish to report an efficient synthesis of this part of the structure, which opens a practical route to sparsomycin (1) and analogs for further biochemical and pharmacological studies.

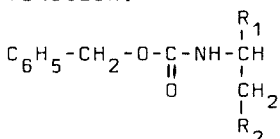


Treatment of N-benzyloxycarbonyl L-cysteine methyl ester (R-configuration) in CH₂Cl₂ with three equivalents of Cl₂ in the presence of Ac₂O⁸ at -10° gave the sulfinylchloride 2 as a white solid⁹. Reaction of 2 with dry CH₂N₂ according to a procedure developed by Venier *et al.*¹⁰ gave the corresponding α-chloro-

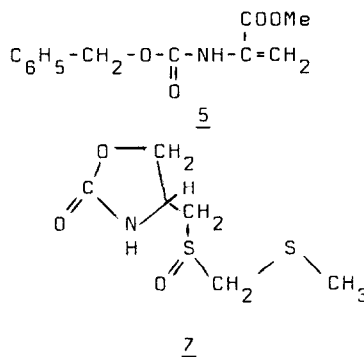
sulfoxide 3⁹. With an undried ethereal CH₂N₂ solution up to 30% of 2 was converted into the sulfinate ester 4. It was found that substitution of Cl in 3 by -SCH₃ had to occur after reduction of the ester function: direct treatment of 3 with CH₃SNa gave the dehydro amino acid derivative 5. The ester function of 3 could be reduced selectively with LiBH₄ in monoglyme yielding the alcohol 6¹¹. Separation by column chromatography on silica gel (Merck 60-H) using CH₂Cl₂/MeOH (94/6, v/v) as eluent gave the R_CS_S/R_CR_S diastereomers of 6

in 34% and 21% overall yield from Cbo-cysteine methylester.

Direct conversion of the alcohol 6 to the desired mono-oxodithioacetal 10 failed; treatment of 6 with CH_3SNa ¹² in CH_3OH at 40° for 24 hrs gave the cyclic urethane 7¹¹ in 30% yield after column chromatography. To circumvent this cyclisation reaction the alcohol function of 6 (mixture of diastereomers) was protected with the tetrahydropyranyl group to yield 8⁹ quantitatively. Treatment of 8 with 1.2 equivalent CH_3SNa in $\text{C}_2\text{H}_5\text{OH}$ for 2 hrs at 60° gave the mono-oxodithioacetal 9. This was converted into the desired compound 10¹¹ by refluxing ethanol in the presence of a trace of HCl . Separation by column chromatography as described above, gave the two possible diastereomers $R_c S_s / R_c R_s$ in 34% and 30% overall yield from 6. On basis of the pmr spectrum [$\delta(\text{CD}_2\text{Cl}_2)$ 7.35 (s, 5H, C_6H_5), 5.82 (br, 1H, NH), 5.10 (s, 2H, $\text{C}_6\text{H}_5\text{CH}_2$), 4.20 (m, 1H, CH), 3.77 (m, 4H, CH_2OH and SCH_2SO), 3.10 (m, 2H, CH-CH_2), 2.29 (s, 3H, SCH_3)] we are inclined to consider the configuration of the major component as enantiomeric with that of sparsomycin (1), which has S configuration at the chiral carbon atom, but unknown configuration at the S(O) function.



- 2 $R_1 = \text{COOMe}, R_2 = \text{S(O)Cl}$
3 $R_1 = \text{COOMe}, R_2 = \text{S(O)CH}_2\text{Cl}$
4 $R_1 = \text{COOMe}, R_2 = \text{SO}_2\text{Me}$
6 $R_1 = \text{CH}_2\text{OH}, R_2 = \text{S(O)CH}_2\text{Cl}$
8 $R_1 = \text{CH}_2\text{OTHP}, R_2 = \text{S(O)CH}_2\text{Cl}$
9 $R_1 = \text{CH}_2\text{OTHP}, R_2 = \text{S(O)CH}_2\text{SCH}_3$
10 $R_1 = \text{CH}_2\text{OH}, R_2 = \text{S(O)CH}_2\text{SCH}_3$



So far, we could not find suitable reaction conditions to remove selectively the N-protecting group. Work is in progress to solve this problem in order to complete the synthesis of sparsomycin (1) and its analogs.

References

1. A.D. Argoudelis and R.R. Herr, *Antimicrob. Ag. Chemother.* 780 (1962).
2. S.P. Owen, A. Dietz and G.W. Camiener, *ibid*, 772 (1962).
3. R.E. Monro and D. Vazques, *J. Mol. Biol.* 28, 161 (1967).
4. C.K. Lee and R. Vince, *J. Med. Chem.* 21, 176 (1978).
5. C.C.L. Lin and R.J. Dubois, *ibid*, 20, 337 (1977).
6. P.F. Wiley and F.A. MacKellar, *J. Amer. Chem. Soc.* 92, 417 (1970); P.F. Wiley and F.A. MacKellar, *J. Org. Chem.* 41, 1858 (1976).
7. H.C.J. Ottenheijm, S.P.J.M. van Nispen and M.J. Sinnige, *Tet. Letters* 1899 (1976).
8. I.B. Douglass and R.V. Norton, *J. Org. Chem.* 33, 2104 (1968).
9. This compound gave satisfactory spectral data.
10. We wish to thank Dr. C.G. Venier for bringing this reaction to our attention; C.G. Venier, H.-H. Hsieh and H.J. Barager, *J. Org. Chem.* 38, 17 (1973).
11. This compound gave satisfactory spectral data and elemental analysis.
12. K. Ogura and G. Tsuchihashi, *Chem. Comm.* 1689 (1970).